

1 **A measurement system for vertical seawater profiles close to the** 2 **air/sea interface**

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10

11 **Abstract**

12 This paper describes a Near Surface Ocean Profiler, which has been designed to precisely
13 measure vertical gradients in the top 10 m of the ocean. Variations in the depth of seawater
14 collection are minimised when using the profiler compared to conventional CTD/rosette
15 deployments. The profiler consists of a remotely operated winch mounted on a tethered yet
16 free floating buoy, which is used to raise and lower a small frame housing sensors and inlet
17 tubing. Seawater at the inlet depth is pumped back to the ship for analysis. The profiler can be
18 used to make continuous vertical profiles or to target a series of discrete depths. The profiler
19 has been successfully deployed during wind speeds up to 10 m s^{-1} and significant wave
20 heights up to 2 m. We demonstrate the potential of the profiler by presenting measured
21 vertical profiles of the trace gases carbon dioxide and dimethylsulfide. Trace gas
22 measurements use an efficient microporous membrane equilibrator to minimise the system
23 response time. The example profiles show vertical gradients in the upper 5 m for temperature,
24 carbon dioxide and dimethylsulfide of $0.15 \text{ }^\circ\text{C}$, $4 \text{ } \mu\text{atm}$ and 0.4 nM respectively.

25

26 **1 Introduction**

27 Exchange between the ocean and atmosphere is an important process for many gases.

28 Important examples include carbon dioxide (CO_2), for which the oceans account for 25% of

1 the sink for anthropogenic emissions (Le Quéré et al., 2016), and dimethylsulfide (DMS),
2 which has an oceanic source and influences cloud properties with implications for the global
3 energy balance (Quinn and Bates, 2011). The magnitude and direction of air/sea gas transfer
4 is typically represented by $\text{Flux} = K\Delta C$ (Liss and Slater, 1974), where ΔC is the concentration
5 difference across the air-sea interface and K is the gas transfer velocity. Direct flux
6 measurements (Bell et al., 2013; Yang et al., 2013; Miller et al., 2010) are only possible for a
7 small number of gases and are not made routinely. Most flux estimates use a wind speed-
8 based parameterisation of K (e.g. Wanninkhof, 2014) coupled with measurements of ΔC .

9 CO_2 is the most well-observed trace gas in the surface ocean, with 14.5 million measurements
10 compiled into a global database, the Surface Ocean CO_2 Atlas (SOCAT),
11 <http://www.socat.info/> (Bakker et al., 2016). Global trace gas databases also exist for gases
12 such as methane and nitrous oxide <https://memento.geomar.de/> (Bange et al., 2009),
13 dimethylsulfide <http://saga.pmel.noaa.gov/dms/> (Lana et al., 2011) and halocarbons
14 <https://halocat.geomar.de/> (Ziska et al., 2013). Accurate estimation of air/sea flux requires
15 concentration measurements that are representative of the interfacial concentration difference.
16 Surface seawater samples are often collected from the underway seawater intake of research
17 vessels, typically at 5-7 m depth. A source of potential error in air/sea flux calculations arises
18 from the assumption of vertical homogeneity within the mixed layer (Robertson and Watson,
19 1992). If vertical concentration gradients exist in the mixed layer, then underway seawater is
20 not representative of the interfacial layer, which could create a global sampling bias (McNeil
21 and Merlivat, 1996).

22 Vertical gradients in trace gas concentrations have been observed under conditions that are
23 favourable for near surface stratification (Royer et al., 2016). At low wind speeds, high solar
24 irradiance can suppress the depth of shear-induced mixing to create a near surface layer
25 several degrees warmer than the water below (Ward et al., 2004; Fairall et al., 1996). Near
26 surface stratification in the marine environment can also be induced by freshwater inputs such
27 as rain (Turk et al., 2010) and riverine discharge. Changes in surface seawater temperature
28 and salinity alter the solubility of dissolved gases and thus the amount available for air/sea
29 exchange (Woolf et al., 2016). Dissolved gases isolated in the upper few metres of the ocean
30 may additionally be modified by physical process such as air/sea exchange and
31 photochemistry. Marine biota confined within the stratified layer (Durham et al., 2009), may

1 also alter trace gas concentrations. For the purposes of this paper, near surface gradients are
2 defined as physical and/or chemical gradients in the upper 10 m of the ocean.

3 Identifying and quantifying near surface gradients in trace gas concentrations is challenging.
4 Ship motion often inhibits near surface measurements made with the standard oceanographic
5 approach of sampling with Niskin bottles mounted on a CTD rosette. Substantial vertical
6 movement of the rosette limits how close to the surface a sample can be taken. For example, a
7 crane arm 4 m above the sea surface and 11 m from the centreline of a ship that is rolling by
8 ± 4 degrees will induce ~ 1.5 m sample depth variation every few seconds. CTD/Niskin bottle
9 sampling requires that the rosette is kept below the sea surface. Sampling within 2 m of the
10 sea surface is often impossible, even under relatively calm conditions.

11 We present a Near Surface Ocean Profiling buoy (NSOP) designed for measuring near surface
12 profiles. The design principles for NSOP were:

13 (1) Platform diameter less than the wavelength of most open ocean waves, allowing it to ride
14 the swell;

15 (2) Short sampling arm close to the sea surface to reduce vertical movements induced by
16 platform motion;

17 (3) Capable of deployment close to the ship (to retrieve water for trace gas analysis), but away
18 from major turbulence and motion due to the ship itself.

19 Example profiles from a cruise on the European continental shelf (*RRS Discovery*, DY033,
20 July 2015) and in the English Channel on board the *RV Plymouth Quest* (part of the Western
21 Channel Observatory, Smyth et al., 2010, April 2014) are discussed.

22

23 **2 Methods**

24 **2.1 Near Surface Ocean Profiler (NSOP) description**

25 NSOP is a repurposed ocean buoy (1.6 m diameter) with a central lifting eyelet (Fig. 1). The
26 top of the buoy is 0.5 m above the sea surface. Mounted on top of the buoy are a line of sight,
27 remotely operated winch (Warrior Winch, model C8000) and a gel battery (Haze, model
28 HZY-S112-230). The winch feeds Kevlar rope through a block and tackle with a 3:1 ratio to
29 reduce rope pay-out speed to ~ 0.05 m s⁻¹. The block and tackle is attached to the end of an

1 outstretched arm 0.25 m from the outer edge of the buoy. The winch line is attached to an
2 open frame (0.35 m diameter, 0.8 m height) with the capacity to house multiple sensors.
3 Desired sampling depth is targeted using knowledge of the winch pay-out speed. Rope pay-
4 out is then timed with a stopwatch. This approach only approximately regulates the sampling
5 depth because: (i) winch pay-out varies slightly depending on the amount of rope on the
6 spool; and (ii) variable horizontal current strength affects the vertical versus horizontal
7 position of the sampling frame. To minimise horizontal movement of the sampling frame we
8 attached a 10 kg weight to the base of the frame.

9 The primary sensor on the sampling frame is a small CTD (Valeport miniCTD) set to sample
10 at a high frequency (>1 Hz). Under calm conditions it is possible to sample as close as 0.1 m
11 from the air/sea interface when the miniCTD and tubing are mounted near the top of the
12 frame. Rougher conditions demand that the frame be kept deeper (~0.5 m) as motion can
13 momentarily bring the sensors and tubing out of the water. An emergency tag line was
14 attached to the sampling frame in case the winch line failed. Seawater for trace gas analysis
15 was pumped back to the ship at 3.5 L min^{-1} through a 50 m PVC hose (0.5 in inner diameter).
16 A heavy duty peristaltic pump (Watson Marlow, model 701IB/R), primed with water from the
17 ships underway supply was used to overcome the large hydraulic head (~4 m). The open end
18 of the tubing was located at the same depth as the miniCTD. Water arriving to the ship's
19 laboratory was divided, with $\sim 3.0 \text{ L min}^{-1}$ for flow-through analysis (e.g. equilibrator for trace
20 gases) and $\sim 0.5 \text{ L min}^{-1}$ for discrete samples (e.g. total alkalinity).

21 We assessed the depth resolution capability of NSOP at a particular depth by looking at
22 pressure variations under calm conditions with a fixed amount of winch rope paid out. In calm
23 to moderate conditions (<2.5 m significant wave height) the amount of vertical movement
24 indicated by the standard deviation (SD) in the depth is $\pm 0.18 \text{ m}$ (Fig. S1, supplementary
25 information). During 4 deployments in rough conditions (>2.5 m significant wave height), the
26 depth variability increased as the sampling frame was lowered (at 5 m, SD was $\pm 0.275 \text{ m}$).

27 **2.2 NSOP deployment**

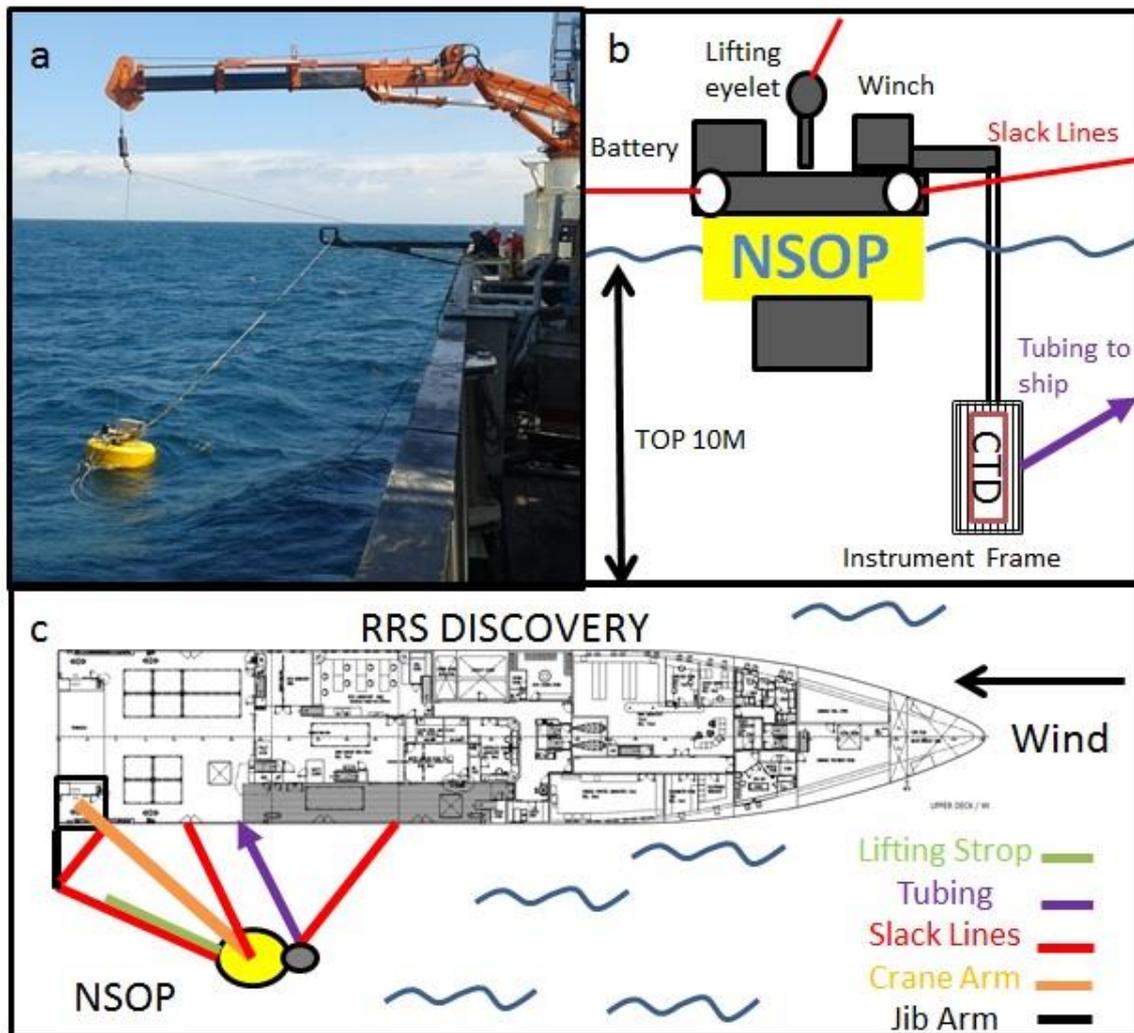
28 On a large research vessel such as the *RRS Discovery*, the deployment and recovery of NSOP
29 requires close coordination between the bridge and 3 personnel on deck. NSOP was always
30 deployed while the ship was on station and not at the same time as other overboard
31 deployments. Ship orientation during deployments was typically with bow into the wind but

1 also accounted for swell and current direction/speed. NSOP was lifted by the aft crane (Fig.
2 1). Once NSOP was lowered to the surface it was detached from the crane via a quick release.
3 Two slack lines were looped through eyelets on the free-floating NSOP to maintain its
4 position close to the ship. A third slack line was connected to the top of the buoy and passed
5 through a block on a fully extended crane arm of 7 m to maintain this distance between NSOP
6 and the ship. The slack lines successfully inhibited the tendency of NSOP to drift horizontally
7 without disrupting its ability to ride the swell. The instrument frame acted like a sea anchor
8 and minimised rotation of NSOP. A 4 m lifting strop used for recovery was connected to the
9 lifting eyelet and loosely lashed to the aft slack line. During retrieval, the slack lines were
10 hauled in and the crane and jib arms brought towards the ship to bring NSOP alongside. The
11 lifting strop was then parted from the slack line and attached to the crane to lift NSOP back on
12 deck. For additional photographs of a NSOP deployment and videos of NSOP during a
13 deployment and in operation see Fig. S2 and videos, supplementary information.

14 Turbulence from the ship's propellers has the potential to mix the water column and destroy
15 any near surface gradients. The ship did not use the aft thrusters whenever conditions were
16 suitable (mild sea state, weak currents and no local hazards). Keeping NSOP away from the
17 ship limited disruption of near surface gradients by the thrusters and reduced the risk of line
18 entanglement in the aft propellers. Our winch did not have a groove bar to feed the rope onto
19 the winch drum, leading to an increased likelihood of snagging during spooling. To minimize
20 snagging, the rope was manually fed onto the winch spool before deployments. Visual
21 monitoring of the NSOP frame, slack lines and winch spool is important during deployment.

22 NSOP has been successfully deployed in 'moderate' sea states up to Beaufort force 5 (~10 m
23 s⁻¹ wind speed and wave heights of ~2.0 m). Deployment length typically varied from 1-3
24 hours.

25



1
 2 **Figure 1:** Different points of view of an NSOP deployment: (a) Image from a deployment on
 3 *RRS Discovery* in May 2015 (Cruise DY030); (b) Schematic cross section of NSOP including
 4 tubing back to ship (purple) and slack lines (red); and (c) Top down schematic from a research
 5 ship including ship orientation. Not to scale.

6
 7 NSOP can be used in two profiling modes: ‘continuous’ and ‘discrete’. Continuous profiling
 8 maximises vertical coverage and involves the winch continuously paying rope in and out at
 9 $\sim 0.05 \text{ m s}^{-1}$. A complete down/up profile to 10 m can be conducted in approximately 7
 10 min. Depth resolution during continuous profiling is determined by the measurement response
 11 time. Instruments with rapid response times such as the miniCTD temperature and
 12 conductivity sensors (0.15 s and 0.09 s) have theoretical depth resolutions of 0.75 cm and
 13 0.45 cm respectively. Actual depth resolution will also be affected by the sampling depth
 14 variability of the NSOP instrument frame. A measurement setup with a longer response time
 15 (such as for seawater CO_2) requires a different approach (see Section 2.5).

1 During discrete profiling, the winch pays out a fixed amount of rope (typically 0.5 m) and the
2 sampling frame is left at a fixed depth. After a fixed sampling period, more rope is paid out.
3 The process is repeated down and then up such that a set of discrete depths are sampled in a
4 'stepped' profile. The discrete profiling depth resolution is determined by the depth
5 fluctuations when sampling at a fixed depth (see Section 2.1). Discrete profiles are a more
6 appropriate approach for measurement systems with a longer response time. A discrete profile
7 with 0.5 m steps down to 5 m and back to the surface using a 2.5 min sampling period takes
8 about an hour. The sampling period at each depth and frequency/distribution of depths within
9 the profile can be adjusted to suit sampling priorities.

10 The maximum deployment time is limited by the capacity of the winch battery. When under
11 no load, the battery allows for approximately 3 hours of operation in the continuous mode.
12 Discrete profiling requires substantially less winch usage such that battery drainage is even
13 less of a concern.

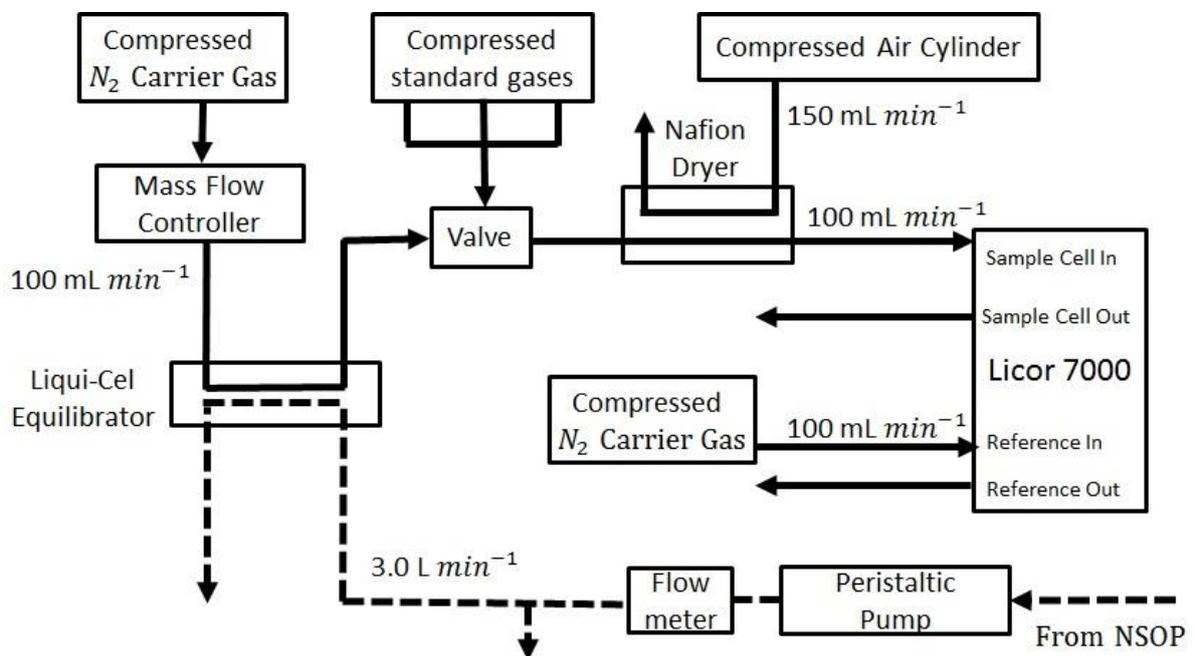
14 **2.3 CO₂ analysis**

15 The CO₂ measurement system (Fig. 2) is a modified version of the system described by
16 (Hales et al., 2004). Seawater from the NSOP inlet was passed through the equilibrator (see
17 Section 2.3.1) at $\sim 3 \text{ L min}^{-1}$ and the flow rate monitored (Cynergy ultrasonic flow meter,
18 model UF25B). A compressed nitrogen gas supply, maintained at a constant flow rate of 100
19 mL min⁻¹ (Bronkhurst mass flow controller, model F-201-CV-100) flows through the
20 equilibrator in the opposite direction to the seawater flow. The gas has high water vapour
21 content after equilibration and is dried (Permapure nafion dryer, model MD-110-48S-4). The
22 dried sample then enters the analytical cell of a NDIR Licor 7000, which is protected with a
23 0.2 μm filter (Pall, Acro 50).

24 CO₂ measurements at atmospheric pressure as recommended by Dickson et al. (2007) were
25 not possible due to the nature of the experimental setup. The continuous gas flow through the
26 system caused a small 0.4 kPa pressure increase in the Licor measurement cell, this was in
27 good agreement with a similar observation by Burke Hales (0.5kpa > ambient pressure;
28 Personal communication). The elevated pressure was taken to be representative of the
29 equilibrator pressure and was used to obtain the partial pressure of CO₂ in the equilibrator
30 ($p\text{CO}_{2(\text{eq})}$).

1 The Licor was calibrated using three CO₂ standard gases before and after each NSOP
 2 deployment. The concentrations of the standard gases (BOC Ltd.) were determined by
 3 referencing against US National Oceanic and Atmospheric Administration certified standards
 4 (244.91, 388.62, 444.40 ppm) in the laboratory. The seawater temperature at the entry and
 5 exit ports of the equilibrator was recorded at 1 Hz (Omega ultra-precise 1/10 DIN immersion
 6 RTD) using stackable microcontrollers (Tinkerforge master brick 2.1 and PTC bricklet).
 7 Equilibrator temperature probes and the miniCTD temperature sensor were calibrated before
 8 and after each cruise against an accurate reference sensor (Fluke, model 5616-12, ±0.011°C)
 9 in a stable water bath (Fluke 7321).

10



11

12 **Figure 2:** CO₂ system schematic. Solid and dashed arrows correspond to gas and water flows
 13 respectively. The Licor reference cell is flushed with equilibrated gas at 100 mL min⁻¹. A
 14 manual selection valve was used to switch between equilibrated gas and the CO₂ standards.

15

16 2.3.1 Equilibrator

17 The showerhead equilibrator is the most commonly-used equilibrator for CO₂ but takes ~100
 18 s to equilibrate (Dickson et al., 2007; Kitidis et al., 2012; Körtzinger et al., 2000; Webb et al.,
 19 2016). This equilibration time is too slow for effective use during NSOP deployments. We
 20 used a polypropylene membrane equilibrator (Liqui-Cel, model 2.5x8) with liquid and gas

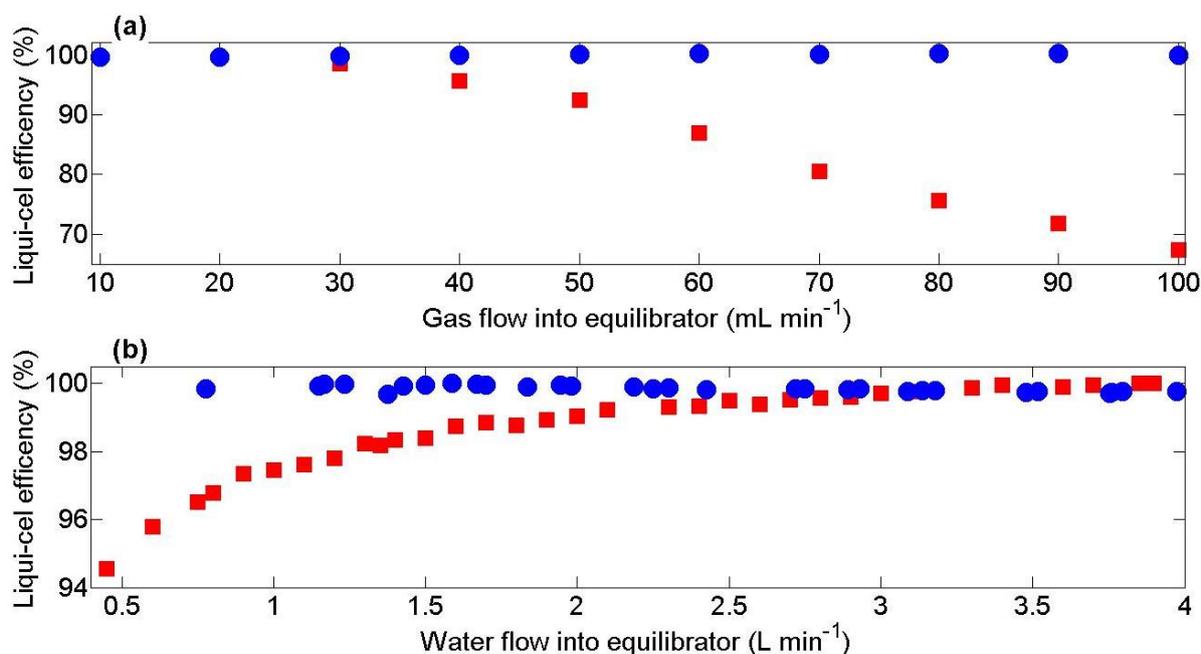
1 volumes of 0.4 L and 0.15 L and a surface area of 1.4 m². Due to its large surface area to
2 volume ratio and membrane porosity (50%), the Liqui-Cel expedites gas transfer and
3 efficiently achieves equilibration (Loose et al., 2009), with a 3 s response time for CO₂ (Hales
4 et al., 2004). Membrane equilibrators have been used by others for trace gas analysis (Hales et
5 al., 2004; Marandino et al., 2009).

6 Fugacity of seawater CO₂ is calculated from the Licor gas phase CO₂ measurement. This
7 approach assumes that the gas phase sample has equilibrated fully with the seawater. We
8 performed equilibration efficiency experiments in a seawater tank using a showerhead
9 equilibrator as a reference. Liqui-Cel equilibration efficiency declined after prolonged
10 exposure to seawater, likely due to biofouling of the membranes. In a fouled equilibrator,
11 equilibration efficiency was a function of the flow rate on both the water and gas side of the
12 membrane. An increased gas flow rate reduces the residence time inside the Liqui-Cel and
13 allows less time to equilibrate (Fig. 3a). Increasing the waterside flow rate moves the gas
14 phase closer to equilibrium because the transfer coefficient in the membrane increases (Fig.
15 3b).

16 Cleaning with an acid - base sequence restored the efficiency of a fouled equilibrator. It was
17 necessary to actively pump chemicals through the Liqui-Cel to achieve a full recovery in
18 efficiency. For more details on cleaning techniques, see supplemental material. Efficiency
19 reductions in membrane equilibrators like the Liqui-Cel have not been reported by previous
20 studies. Some authors have used 5-50 µm filters to minimise biofouling (Hales et al., 2004)
21 but this was not possible with the NSOP experimental design. If filtering seawater is not
22 possible, we recommend flushing with freshwater after use, regular cleaning of the Liqui-Cel
23 and daily tests to quantify equilibration efficiency. Trace gas measurement systems that use
24 an internal liquid phase standard (e.g. dimethylsulfide, Section 2.4) account for any changes
25 in equilibrator efficiency.

26

27



1
2 **Figure 3:** Liqui-Cel CO₂ equilibration efficiency (Liqui-Cel mixing ratio / showerhead
3 mixing ratio) for: (a) changing gas flow at a fixed water flow rate of 4 L min⁻¹; and (b)
4 changing water flow at a fixed gas flow of 100 mL min⁻¹. Blue = unfouled equilibrator. Red =
5 fouled equilibrator.

6 2.4 DMS analysis

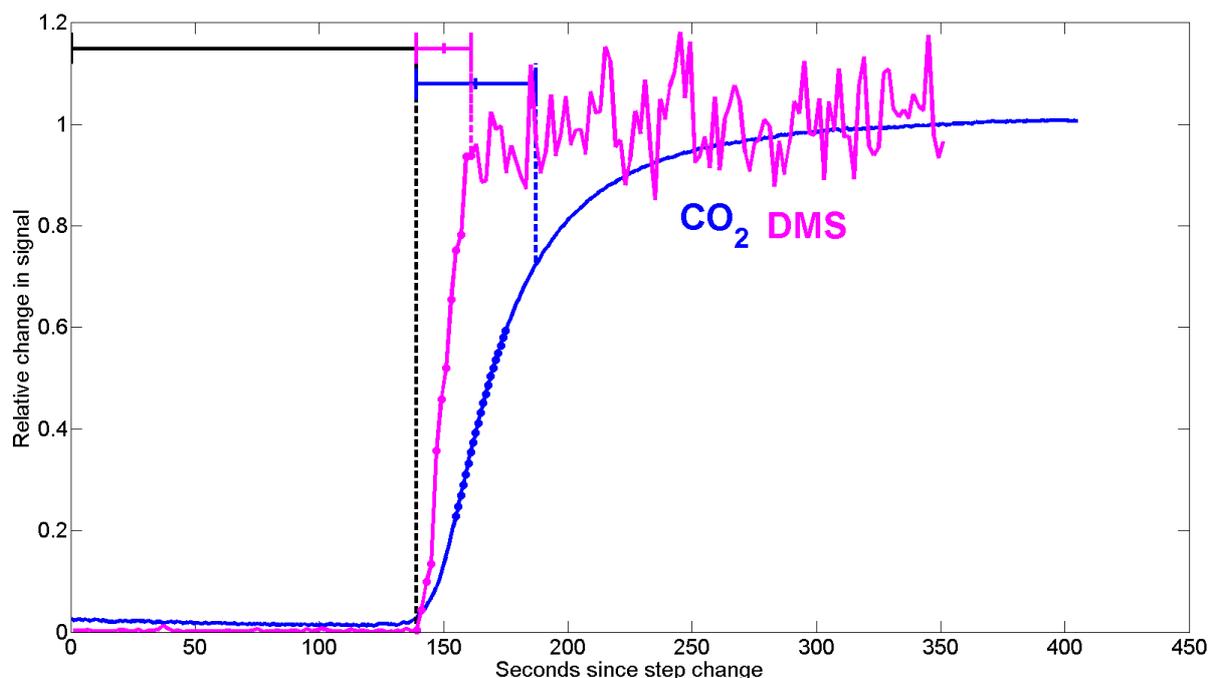
7 DMS was measured with Atmospheric Pressure-Chemical Ionisation Mass Spectrometry
8 (API-CIMS), using a system modified following Saltzman et al. (2009). Measurements were
9 calibrated using an isotopic liquid standard of tri-deuterated DMS (see Bell et al., 2013 for
10 details). Isotopic standard was injected at 120 $\mu\text{L min}^{-1}$ into the 3 L min⁻¹ seawater flow from
11 NSOP before it entered the Liqui-Cel equilibrator. Compressed nitrogen gas was passed
12 through the equilibrator in the counter direction to the seawater flow at 1 L min⁻¹. The use of
13 an internal standard meant that any incomplete equilibration of the ambient non-isotopic DMS
14 was also true for the isotope. The gas stream exited the equilibrator and was dried (Permapure
15 nafion dryer, model MD-110-48S-4) before entering the mass spectrometer for analysis. DMS
16 was detected at m/z (mass/charge) 63 and the isotopic standard detected at m/z 66. The
17 concentration of DMS was calculated using the ion signals and relevant flow rates (Bell et al.,
18 2015). This approach has been shown to compare well with other analytical techniques for
19 DMS (Royer et al., 2014; Walker et al., 2016).

1 **2.5 NSOP delay and response time**

2 We used different approaches to assess the delay between instantaneous miniCTD
3 measurements and water arriving to the ship for analysis. The delay between seawater
4 entering the inlet and reaching the equilibrator was calculated as 114 s using the internal
5 volume of NSOP tubing (0.5 in inner diameter, 54 m length) and a seawater flow rate of 4.15
6 L min⁻¹. Delay correlation analysis between the NSOP miniCTD temperature sensor and a
7 second sensor positioned at the entrance to the equilibrator gives a similar delay of 112 s.
8 Note that the total delay of the system is greater because it also includes the time that
9 equilibrated gas takes to reach the Licor. We determined the total delay by quickly
10 transferring the seawater inlet quickly between two buckets with distinctly different CO₂
11 concentrations and timing how long it took for the signal to be detected by the Licor (139 s;
12 Fig. 4).

13 The response time of the NSOP setup was determined by simulating step changes in gas
14 concentrations. A model fit to the exponential change in signal was used to estimate the
15 response time (Fig. 4). We estimate the system response time (e-folding time) for CO₂ as 24 s,
16 which is slightly faster than the 34 s reported by Webb et al. (2016). The e-folding time in the
17 DMS signal is estimated as 11 s, which is consistent with the rapid gas flow rate through the
18 analytical system.

19 Continuous profiling with the CO₂ system and a 24 s response time yields a depth resolution
20 of 1.2 m, which is greater than the required resolution to assess near surface gradients. DMS
21 has a faster response time than CO₂, but in continuous profiling mode this only translates to a
22 depth resolution of 0.6 m, slightly less than the 1.2-2 m reported by (Royer et al., 2014). A
23 depth resolution of < 0.5 m was desired to capture upper ocean vertical gradients in CO₂ and
24 DMS so NSOP was operated in discrete profiling mode.



1
 2 **Figure 4:** Instrument responses to step changes in seawater CO₂ (blue) and DMS (magenta).
 3 Step changes from 350 to 400 μatm for CO₂ and 0 to 2 nmol L⁻¹ for DMS have been scaled
 4 down so that the initial and end concentrations are between 0 and 1. Time is referenced
 5 against the point when the step change was initiated. The response is seen in both instruments
 6 after a delay of 138 s (black dashed line). Two e-foldings are indicated by vertical dashed
 7 lines for CO₂ (blue) and DMS (magenta). The data points marked by circles were used to
 8 make an exponential fit to the data to determine the response time (Sect 2.5).

10 2.6 Data processing

11 During discrete profiling, distinct sample depths were identified from the rapid changes in
 12 pressure during depth transitions. Data were binned into discrete depth bins using CTD
 13 pressure measurements. Trace gas data were assigned to depth bins after adjusting for the
 14 calculated transit time through the NSOP tubing (Section 2.5). CO₂ data from the beginning
 15 (2 e-foldings + 15 s buffer = 63 s) and end (15 s buffer) of each depth bin was excluded from
 16 analysis to account for the response time of the system and the transition time between sample
 17 depths. The same approach was taken for DMS, where the faster response time resulted in a
 18 smaller portion of data excluded at the beginning of each depth bin (2 e-foldings + 15 s buffer
 19 = 37 s).

20 The CO₂ mixing ratio (xCO₂) measured in the Licor is converted to equilibrator fugacity
 21 (fCO_{2(eq)}) using calibration standards, *in situ* seawater salinity, and the pressure and

1 temperature in the equilibrator (SOP 5# Underway pCO₂ Dickson et al., 2007). Vertical
2 profiles of seawater CO₂ fugacity (fCO_{2(sw)}) are calculated using average equilibrator fugacity
3 (fCO_{2(eq)}), equilibrator temperature (T_(eq)) and *in situ* seawater temperature (T_(sw)) at each
4 depth (Takahashi et al., 1993).

5 **2.7 Seawater sample collection using NSOP**

6 The NSOP setup enables vertical profiles of discrete seawater samples to be collected from
7 upstream of the equilibrator, with a split in the tubing diverting ~0.5 L min⁻¹ into a sink. For
8 example, discrete seawater samples (250 ml) have been successfully collected and analysed
9 for Total Alkalinity (TA). Samples were collected and poisoned following best practice
10 recommendations (SOP#1, (Dickson et al., 2007). Bottle filling plus 1 overfill took ~60 s.
11 Start and end times were recorded so that collection depth could be retrospectively
12 determined from the CTD pressure data. Analytical methods and an example depth profile are
13 provided in the supplementary information.

14

15 **3 Field Measurements / Observations**

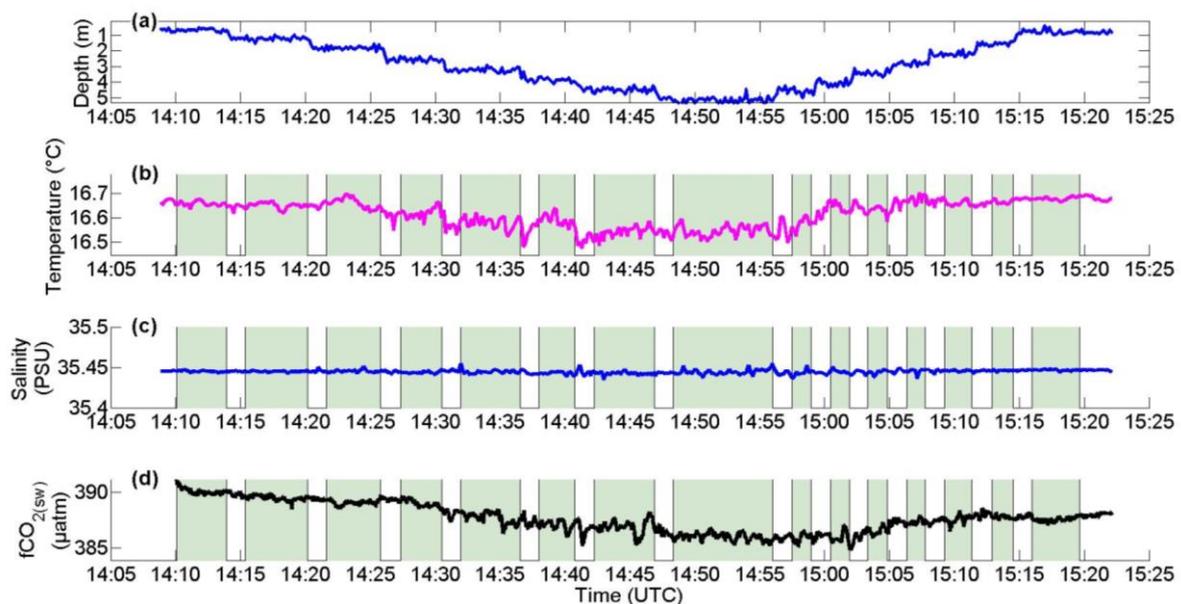
16 Presented below are example profiles collected using NSOP. The first deployment was in the
17 open ocean (July 30th 2015, Central Celtic Sea; 49.4213°N, -8.5783°E) from the *RRS*
18 *Discovery* (100 m length, 6.5 m draught). The second deployment was in coastal waters (15th
19 April 2014, Plymouth Sound; 50.348°N, -4.126°E) from the *RV Plymouth Quest* (20 m
20 length, 3 m draught). A map of deployment sites is supplied in the supplementary
21 information.

22 **3.1 Open ocean deployment**

23 NSOP was deployed at 14:05 (UTC) on 30th July 2015. During the 6 hours preceding
24 deployment, the ship was on station and encountered persistently strong solar radiance (> 600
25 W m⁻²), mild winds (< 6 m s⁻¹) and calm sea state (significant wave height < 1.6 m). This
26 combination of low wind speeds and high irradiance (Fig. S5, supplementary information) is
27 favourable for near surface stratification (Donlon et al., 2002).

28

1 Fig. 5 presents the time series data collected by NSOP for depth, temperature, salinity and
 2 $f\text{CO}_{2(\text{sw})}$. Discrete profiling began at 14:05 hrs (UTC) at 0.7 m depth, which was as close to
 3 the surface as the frame could be located without the possibility of breaking the surface.
 4 Depth bins were identified based on rapid depth transitions (Fig. 5a). Bottles were filled for
 5 discrete samples during the downcast. Profiling lasted 75 minutes and finished back at the
 6 surface at 15:20 hrs (UTC). Seawater temperature was 16.61 ± 0.06 °C. At 14:20 hrs (UTC)
 7 $f\text{CO}_{2(\text{atm})}$ was 398 μatm and $f\text{CO}_{2(\text{sw})}$ was 389 μatm at 0.67 m meaning the ocean was
 8 undersaturated with respect to the atmosphere. The temperature and seawater CO_2 were the
 9 expected magnitude for summer in the Celtic Sea (Frankignoulle and Borges, 2001). Salinity
 10 was homogeneous throughout the NSOP deployment, only varying by ± 0.004 .



11
 12 **Figure 5:** Time series measurements made during an NSOP deployment in the Celtic Sea on
 13 30th July 2015. Data are 1 Hz depth (a), seawater temperature (b), salinity (c) and $f\text{CO}_{2(\text{sw})}$ (d).
 14 Data used for depth bin analysis (Section 2.6) is identified by a shaded background.

15
 16 Depth-binned salinity and temperature data did not show any significant variability (Fig. 6a).
 17 A slight temperature gradient was observed, with 0.15 °C difference between 5 m and the
 18 surface and a fairly constant reduction with depth (0.03 °C per metre). The temperature profile
 19 was similar for down and up casts, although some continued warming of surface waters was
 20 evident in the up cast. The temperature measured by NSOP at 5.15 m depth agrees well with
 21 the coincident temperature measured by the bow thermistor at 5.5 m (< 0.02 °C difference)

1 (Fig. 6c). There is no evidence that the ship's thrusters/propellers disrupted the near surface
2 gradients.

3 We compare the NSOP temperature profile with thermistor readings from a series of Sea-Bird
4 Scientific (SBE 56) sensors (0.3, 0.6, 1.5, 3.5 and 7 m depth) mounted on a nearby
5 temperature chain moored ~2.8 km away (49.403°N, -8.606°E) from the deployment site .

6 The vertical profile implied by the NSOP deployment agrees with the mooring data (Fig. 6c),
7 and corroborates the warming of the upper few metres of the ocean observed during the

8 deployment. The agreement between these independent datasets suggests that it is unlikely

9 that NSOP caused any significant localized warming of surface waters. The mean difference
10 between NSOP temperature from discrete depths and the mooring sensors is 0.02°C. The

11 surface data from the NSOP up cast show less agreement with the mooring, with NSOP

12 temperatures ~0.05 °C lower than the 0.3 m and 0.6 m mooring sensors. During the profile the

13 ship drifted ~1 km from the start position of the profile and a further 0.2 km from the

14 mooring. The small offset between the NSOP surface temperatures and the mooring may be

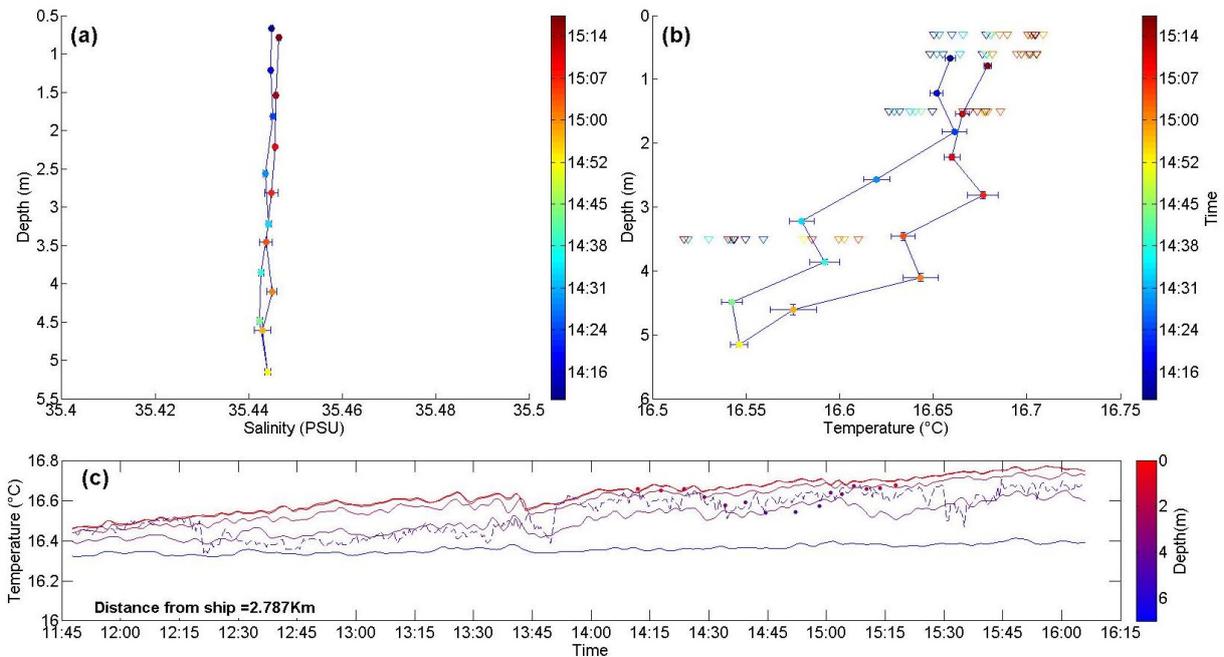
15 driven by horizontal variability between the deployment and mooring locations. It is also

16 possible that turbulence mixed warm surface waters down into cooler sub-surface layers.

17 Turbulence could have been generated around the NSOP sampling frame or by an increase in

18 wave-driven mixing when the significant wave height increased at ~15:00 hrs UTC (Fig.

19 S4a).

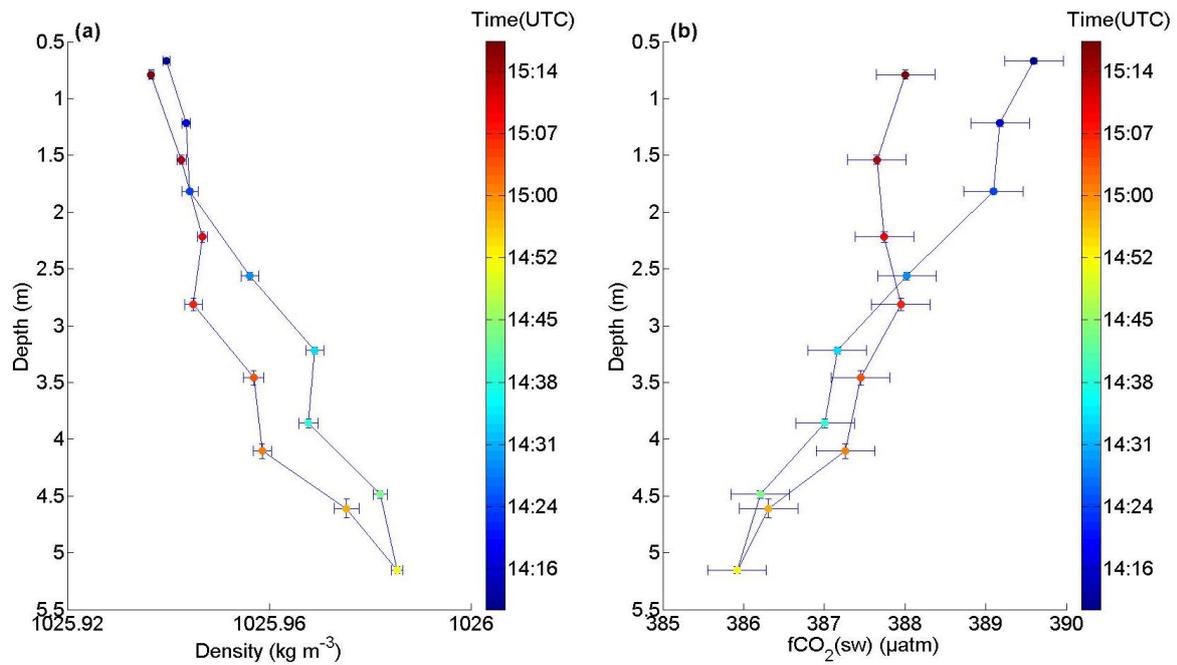


20

1 **Figure 6:** Salinity and temperature in the Central Celtic Sea on 30th July 2015. NSOP profiles
2 of salinity (a) and temperature (b) were derived using depth bins as described in Section 2.6.
3 Data points are coloured by sampling time. Vertical and horizontal error bars show two
4 standard errors of the mean in each depth bin. Coloured triangles in (B) are time-averaged
5 temperature for four depths (0.3, 0.6, 1.5 and 3.5 m) at the nearby Central Celtic Sea
6 temperature mooring (49.403°N, -8.606°E). (c) Timeseries of temperature at the mooring.
7 Timeseries of temperature at depths (0.3, 0.6, 1.5 and 3.5 m) are solid lines whereas the
8 dashed line is the underway temperature at 5.5 m from *RRS Discovery* (located 2.8 km from
9 the mooring). The mooring and underway temperatures are coloured according to their sample
10 depth, where red is the air/sea interface. The circles are binned temperature data from NSOP
11 which have also been coloured to reflect the depth of collection.

12

13 Seawater density (Fig. 7a) was calculated using the salinity and temperature profile data (Fig.
14 6a & 6b) and the 1983 Unesco equation of state (Millero and Poisson, 1981). As expected with
15 little variation in the salinity, changes in the density profile are dominated by temperature.
16 The down and up casts for CO₂ show excellent agreement below 2.5 m. Surface water (< 2 m)
17 CO₂ is 2-4 μ atm higher than at 5 m (Fig. 7b). Elevated surface CO₂ could be explained by a
18 sustained flux from the atmosphere into a near surface stratified layer with inhibited deep
19 water exchange. Under this assumption a vertical gradient in seawater CO₂ would need to be
20 established shortly after the temperature gradient. A paired t-test showed that the CO₂
21 measured in the surface bins on the downcast and upcast are were significantly different (p
22 = <0.001). The deepening of the surface stratified layer could explain the more homogeneous
23 CO₂ during the upcast. It is worth noting that in addition to physical processes, plankton
24 trapped within the surface layer could also modify the surface CO₂. Trace gas concentrations
25 may also be different in the sea surface microlayer but sampling that close to the surface is
26 beyond the capabilities of NSOP. Complimentary measurements of the sea surface microlayer
27 could be made using other state of the art purpose built sampling platforms such as the Sea
28 Surface Scanner (Ribas-Ribas et al., 2017).



1
 2 **Figure 7:** NSOP density (a) and fCO₂(sw) (b) profiles from the Celtic Sea on 30th July 2015.
 3 Data points are coloured by sample time. Vertical error bars correspond to two standard
 4 errors of the mean in each depth bin. The horizontal error bars in (a) are two standard errors of
 5 the mean, whereas in (b) they are the propagated error from the the binned measurements used
 6 to calculate fCO₂(sw).

7
 8 To assess measurement accuracy the NSOP Liqui-Cel CO₂ system was compared against an
 9 independent CO₂ system that had a showerhead equilibrator coupled to the ship's seawater
 10 supply pumped from 5.5 m below the sea surface (Hardman-Mountford et al., 2008; Kitidis et
 11 al., 2012). Technical issues meant that the underway CO₂ system installed on the *RRS*
 12 *Discovery* was not functioning during the deployment detailed above. However during a
 13 deployment on the 19th July 2015, the fCO₂(sw) measured by NSOP at 5 m agreed well with
 14 independent measurements from the underway system, difference = 1.7 +/- 4.18 µatm. The
 15 agreement between the two systems is in line with previous intercomparisons (Ribas-Ribas et
 16 al., 2014; Körtzinger et al., 2000).

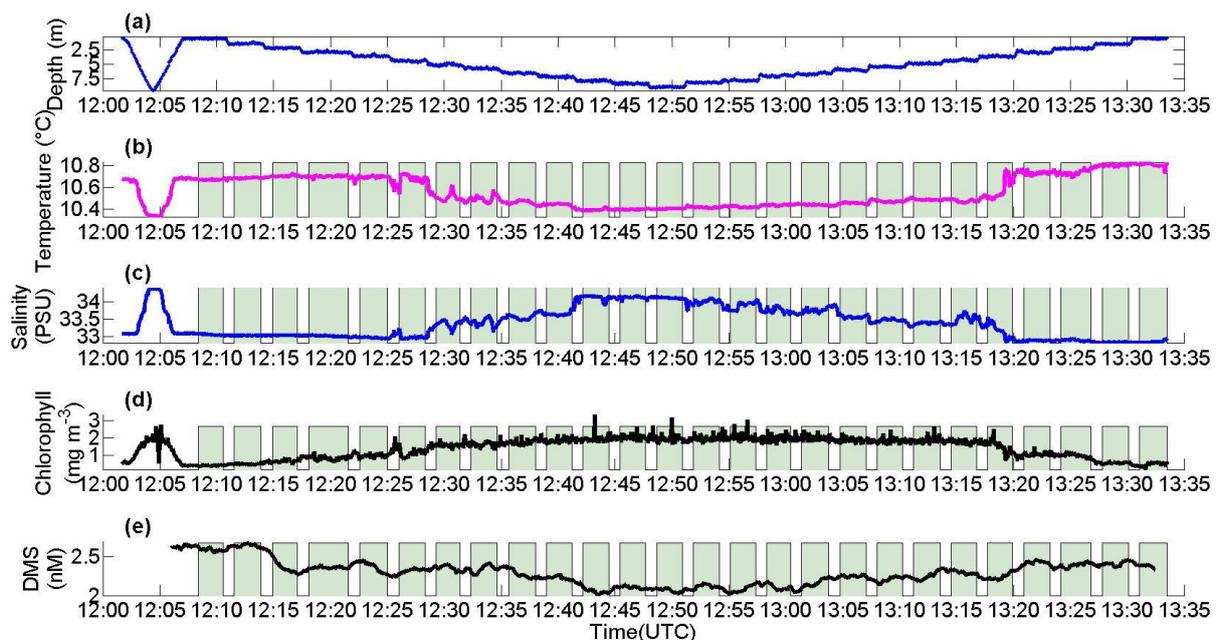
17

18 3.2 Coastal deployment

19 DMS profiles were collected on a small research vessel on 15th April 2014. NSOP was
 20 deployed within the Plymouth Sound at 12:00 hrs UTC and recovered 95 minutes later (Fig.
 21 8). In the sheltered environment behind the breakwater the standard deviation in depth was

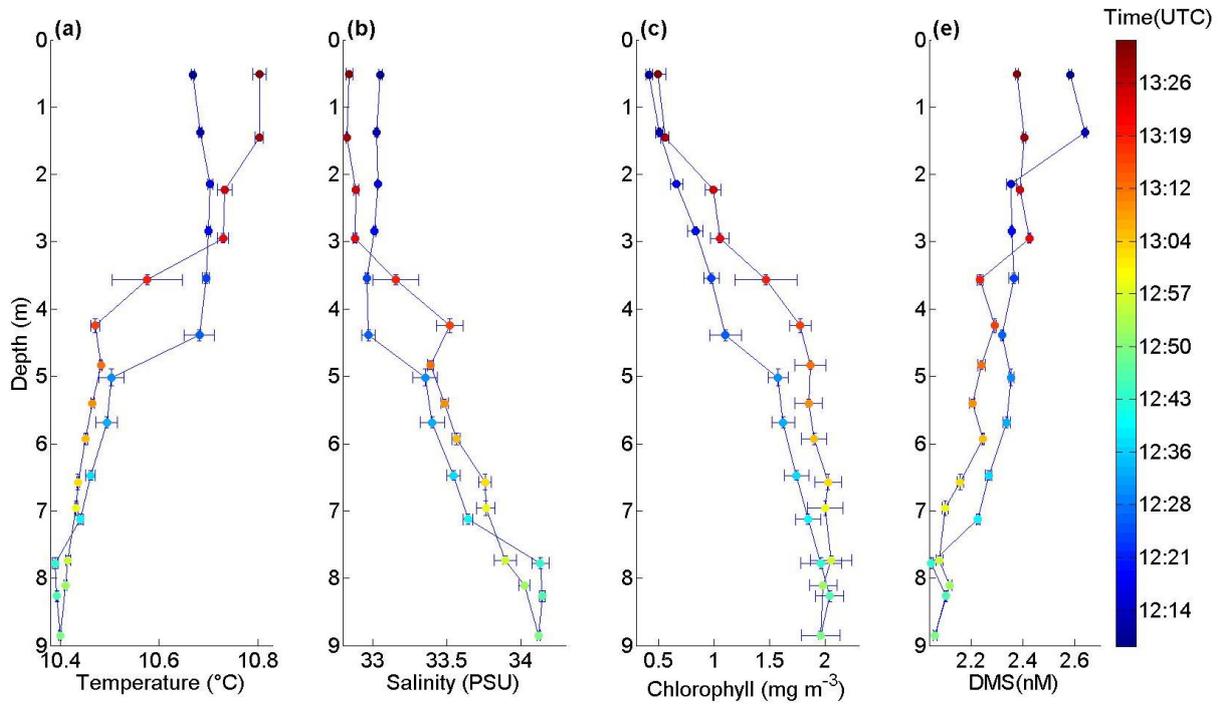
1 ± 0.10 m, smaller than observed during open ocean profiles. Seawater temperature and salinity
2 demonstrate clear structure, with lower temperatures and higher salinities associated with sub-
3 surface water. Two river estuaries (Plym and Tamar) converge and flow out to the open ocean
4 through the Plymouth Sound. We likely observed a freshwater surface lens that was protected
5 from wave-driven mixing and had been warmed over the course of the day. We used a
6 different miniCTD during this deployment and were thus also able to collect fluorescence
7 data (Fig. 8d).

8 Temperature profiles (Fig. 9a) show a sharp discontinuity in the downcast at ~ 5 m whereas in
9 the upcast the thermocline had shoaled to ~ 3.5 m. The salinity profiles suggest similar mixing
10 depths to the temperature profiles, with lower salinity water at the surface (Fig. 9b). The
11 increase in fluorescence with depth (Fig. 9c) is either due to reductions in chlorophyll
12 concentration close to the sea surface or because of quenching of the phytoplankton
13 photosynthetic apparatus, which is often observed in surface waters that experience strong
14 irradiance (Sackmann et al., 2008). DMS concentrations reduce steadily with depth (Fig. 9d),
15 which is likely explained by changes in DMS production and consumption rates by the
16 biological community (Galí et al., 2013). The DMS profiles from the upcast and the downcast
17 are very similar, with the largest difference at the very surface. A large difference in the
18 surface-most data point can also be seen in the temperature data, and may reflect mixing with
19 sub-surface waters due to the motion of NSOP or short time-scale variations in the physical
20 environment.



21

1 **Figure 8:** Time series measurements during an NSOP deployment in Plymouth Sound on 15th
 2 April 2014: depth (a), temperature (b), salinity (c), chlorophyll fluorescence (d) and DMS_(sw)
 3 (e). Data used for depth bin analysis (Section 2.6) is identified by a shaded background. The
 4 beginning of the time series is an example off a continuous profile (see Section 2.2).
 5



6
 7 **Figure 9:** NSOP profiles collected in Plymouth Sound on 15th April 2014: temperature (a),
 8 salinity (b), chlorophyll fluorescence (c), and DMS_(sw) (d). Data are coloured by sample time.
 9 Vertical and horizontal error bars are two standard errors of the mean (SEM) in each depth
 10 bin.
 11

12 4 Summary

13 This paper describes a Near Surface Ocean Profiler (NSOP) designed to measure vertical
 14 trace gas profiles near the air-sea interface. NSOP is unique in approach as its sampling frame
 15 is lowered from a buoy that rides the ocean swell, reducing relative motion of the frame and
 16 hence fluctuations in sampling depth. The NSOP design facilitates near surface (< 0.5 m)
 17 sampling, significantly improving the capability to resolve vertical gradients. Other benefits
 18 include the ability to sample away from ship-driven turbulence and the flexibility to make a
 19 large range of near surface measurements. The NSOP sampling frame houses the miniCTD
 20 and also has the capacity to incorporate additional sensors (e.g. turbulence, dissolved oxygen
 21 and other measures of phytoplankton abundance and photosynthetic health). The ability to

1 collect water from discrete depths facilitates the collection of near surface samples that
2 require additional processing or take longer to analyse (e.g. TA, dissolved inorganic carbon,
3 nutrients, the DMS-precursor DMSP, dissolved organic carbon). NSOP is highly versatile and
4 can be used for continuous or discrete profiling. Further development could adjust winch pay
5 out speed and enable continuous, high resolution depth profiles for slower response time
6 measurements (e.g. $f\text{CO}_{2(\text{sw})}$).

7 Near surface stratification in the upper few metres of the ocean due to temperature and
8 salinity gradients is a well-documented phenomenon. The presence or absence of chemical
9 and biological gradients within near surface stratified layers has been difficult to assess.
10 NSOP is a platform with the capability to successfully resolve gradients in these near surface
11 layers. The data presented in this paper demonstrate that near surface gradients in trace gases
12 can lead to substantially different fluxes depending upon the seawater depth that is used to
13 calculate the flux. Assuming that the effect of temperature and salinity gradients on the flux
14 can be accounted for using remote sensing methods (e.g. Shutler et al., 2016), then the change
15 in flux is directly proportional to the change in ΔC . In the case of the coastal DMS profile, a
16 higher concentration (2.58 ± 0.02 nM) was observed 0.5 m below the sea surface compared to
17 concentrations at 5 m (2.36 ± 0.03 nM). Assuming that the atmospheric concentration of DMS
18 was negligible (a typical approach for DMS fluxes, (see Lana et al., 2011) , computing the
19 flux with the 5 m waterside concentration instead of the 0.5 m waterside concentration means
20 the flux is underestimated by 9.3%. . In the case of the Celtic Sea CO_2 profile, the
21 concentration at 0.5 m (389.60 ± 0.36 μatm) was higher than at 5 m (385.92 ± 0.36 μatm). The
22 atmospheric CO_2 concentration was 398.1 ± 0.3 μatm , which means that the surface water was
23 less undersaturated than implied by the seawater concentration at 5 m. Using the 5 m
24 waterside CO_2 concentration leads to an overestimation of the ΔC and flux by 43.5%
25 compared to using the 0.5 m waterside CO_2 concentration. The magnitudes of these
26 concentration gradients are significant. However, such gradients (in magnitude and direction)
27 do not persist for all hours of the day, under different environmental conditions and in all
28 regions of the global ocean. A subsequent publication will discuss NSOP data collected
29 during four cruises as well as the wider prevalence and implications of near surface CO_2
30 gradients

1 **Acknowledgements**

2 We thank the captains and crews of the *RV Plymouth Quest* and *RRS Discovery* for their
3 assistance with deploying NSOP, Christopher Balfour and Dave Sivyer for maintenance of
4 the Central Celtic Sea mooring near surface temperature sensors, Vassilis Kitidis for
5 supplying underway CO₂ data and Burke Hales for advice concerning Liqui-Cel CO₂
6 measurements. This research was made possible by PML internal funding, a NERC funded
7 studentship (NE/L000075/1), temperature sensors on the Central Celtic Sea mooring
8 (NE/K002058/1) and by the NERC Shelf Sea Biogeochemistry pelagic research programme
9 (NE/K002007/1). The *RRS Discovery* underway data was supplied by the Natural
10 Environment Research Council.

11

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